## Physics 2048, General Physics A Prof. Stephen Hill, Course Leader

## An Introduction to Mechanics, Waves and Thermodynamics

Calculus and Trigonometry will be used (MAC 2311 a pre-requisite)


## MAGLAB

## In collaboration with:

Dr. Y. Hori, KEN507, yhori@fsu.edu
Prof. H.-K. Ng, KEN416, hkng@fsu.edu
Prof. S. Tabor, KEN213, tabor@nucmar.physics.fsu.edu


My coordinates:
Most of the time, I reside at the MagLab on the SW Campus (644-1647)
I will usually be on the main campus Tue/Thu mornings, KEN310 (645-8793)

## fs.magnet.fsu.edu/~shill/



## Newton's laws:

From the Gatorade ad:
A. "What makes bodies in motion remain in motion?"
B. "What makes what goes up, stay up?"
C. "What makes what goes down, get back up?"

One of these statements may be attributed to Newton
**The answer is $A^{* *}$
Newton's first law (Law of inertia): A body at rest remains at rest, and a body in motion will remain in motion at a constant velocity unless acted upon by an external force.

## Standing the test of time

- The classical laws of physics, including mechanics and thermodynamics, have been rigorously studied through centuries of experiment.
- This is the so-called "scientific method."
- During this course, we will explore many (though by no means all**) of the classical laws of physics using controlled experimental demonstrations.
- We will compare the results with simple mathematically based theoretical principles.
- We will then use these theoretical principles to solve a wide range of problems.
**PHY2049 will introduce you to more of these laws.


## What do physicists do?

At the end of the 19th century, A. A. Michelson (very famous physicist) stated that "all of the grand underlying physical principles had been firmly established."

Then came two revolutions:

- Relativity
concepts of space and time change at large relative velocities
-Quantum mechanics
concept of matter changes on small length scales
-Classical laws of mechanics break down in these limits, and much remains to be discovered
These are the things physicists study today through experiment and theory, just as physicists have done through the ages.


## Physics, the 21st century, \& you

A technological revolution
-20th century microelectronics \& the computer revolution
-Now we have "nanoscience" (and molecular sciences)
Devices/molecules which are made up from just a few atoms
1 nanometer $=1$ meter $/ 1,000,000,000$
1 millionth of the diameter of a grain of rice

- Nanoscience is revolutionizing electronic and mechanical engineering, biology, chemistry and medicine
- Physics is playing a larger and larger role in all of these fields

However, before you can tackle these modern subjects, you have to have a fundamental grasp of the underlying classical laws of physics

# Ch.1: International System of Units 

## Système International (SI) d'Unités in French

 a.k.a. metric or SI units- Scientists measure quantities through comparisons with standards.
-Every measured quantity has an associated unit.
-The important thing is to define sensible and practical "units" and "standards" that scientists everywhere can agree upon. (e.g. Rocket Scientists)


## Ch.1: International System of Units

## Système International (SI) d'Unités in French

 a.k.a. metric or SI units- Even though there exist an essentially infinite number of different physical quantities, we need no more than seven base units/quantities from which all others can be derived.
In 1971, the 14th General Conference on Weights and Measures picked these seven base quantities for the SI, or metric system.
-In Mechanics, we really only need three of these base units (see table)
- In Thermodynamics, we need two more
Temperature in kelvin (K) Quantity in moles (mol)

| Quantity | Unit Name | Unit Symbol |
| :--- | :---: | :---: |
| Length | meter | m |
| Time | second | s |
| Mass | kilogram | kg |

## Prefixes for SI Units

| Factor | Prefix ${ }^{a}$ | Symbol |  |
| :---: | :---: | :---: | :---: |
| $10^{24}$ | yotta- | Y | Sclentitic notation: |
| $10^{21}$ | zetta- | Z |  |
| $10^{18}$ | exa- | E | 3560000001109 |
| $10^{15}$ | peta- | P |  |
| $10^{12}$ | tera- | T | $0.000000492 \mathrm{~s}=4.92 \times 10^{-7} \mathrm{~s}$ |
| $10^{9}$ | giga- | G |  |
| $10^{6}$ | mega- | M | On LONCAPA $=4.92 E-7 \mathrm{~s}$ |
| $10^{3}$ | kilo- | k | On Morr calcilator. |
| $10^{2}$ | hecto- | h | O your caicuiator |
| $10^{1}$ | deka- | da | $492 \times 10-7=492 \mathrm{E}-7$ or $492-07$ |
| $10^{-1}$ | deci- | d |  |
| $10^{-2}$ | centi- | c |  |
| $10^{-3}$ | milli- | m |  |
| $10^{-6}$ | micro- | $\mu$ | Prefixes (also works in LONCAPA). |
| $10^{-9}$ | nano- | n | $5 \times 10^{9}$ watts $=5$ gigawatts $=1 \mathrm{GW}$ |
| $10^{-12}$ | pico- | p |  |
| $10^{-15}$ | femto- | f | $2 \times 10^{-9} \mathrm{~s}=2 \text { nanoseconds }=2 \mathrm{~ns}$ |
| $10^{-18}$ | atto- | a |  |
| $10^{-21}$ | zepto- | Z |  |
| $10^{-24}$ | yocto- | y |  |

## Orders of magnitude

| Measurement | Length in Met |
| :--- | :---: |
| Distance to the first galaxies formed | $2 \times 10^{26}$ |
| Distance to the Andromeda galaxy | $2 \times 10^{22}$ |
| Distance to the nearest star (Proxima Centauri) | $4 \times 10^{16}$ |
| Distance to Pluto | $6 \times 10^{12}$ |
| Radius of Earth | $6 \times 10^{6}$ |
| Height of Mt. Everest | $9 \times 10^{3}$ |
| Thickness of this page | $1 \times 10^{-4}$ |
| Length of a typical virus | $1 \times 10^{-8}$ |
| Radius of a hydrogen atom | $5 \times 10^{-11}$ |
| Radius of a proton | $1 \times 10^{-15}$ |

## Changing units

Chain-link conversion - an example:

$$
1 \text { minute }=60 \text { seconds }
$$

$$
\text { therefore } \quad \frac{1 \mathrm{~min}}{60 \mathrm{sec}}=1 \quad \text { or } \quad \frac{60 \mathrm{sec}}{1 \mathrm{~min}}=1
$$

Note: this does not imply $60=1$, or $1 / 60=1$ !
$2 \min =(2 \mathrm{~min}) \times(1)=(2 \mathrm{~min}) \times\left(\frac{60 \mathrm{~s}}{1 \mathrm{~min}}\right)=120 \mathrm{~s}$

## Conversion Factors

| Equivalent Measures of Length |  |
| :---: | :---: |
| 1 meter (m) | 39.37 inches (in.) |
| 1 centimeter (cm) | 0.39 in. |
| 1 millimeter (mm) | 0.039 in. |
| 1 yard (yd) | 91.44 centimeters (cm) |
| 1 foot (ft) | 30.48 cm |
| 1 inch (in.) | 2.54 cm |
| Household Measures (Approximate) |  |
| 1 drop | $1 / 20 \mathrm{~mL}$ |
| 1 teaspoon | 5 mL |
| 1 tablespoon | 15 mL |
| 1 cup | 250 mL |
| Weight and Apothecaries' Equivalents |  |
| 1 milligram (mg) | 1/65 grain (1/60) |
| 1 gram (g) | 15.43 grains (15) |

Etc., etc., etc.......

## Resources at your fingertips...



The LearningOnline Network with CAPA


## Quiz \#1

Which of the following have been used as the standard for the unit of length corresponding to 1 meter?
a) One ten millionth of the distance from the North pole to the equator.
b) The distance between two fine lines engraved near the ends of a platinum-iridium bar.
c) 1650763.73 wavelengths of a particular orange-red light emitted by atoms of krypton-86 ( ${ }^{86} \mathrm{Kr}$ ).
d) The length of the path traveled by light in a vacuum during a time interval of $1 / 299792458$ of a second.

## Length

1792: French established a new system of weights and measures

$$
1 \mathrm{~m}=\frac{\text { distance from } N \text {. pole to equator }}{\text { ten-million }}
$$

- Then, in the 1870s:
$1 \mathrm{~m}=$ distance between fine lines on Pt -Ir bar Accurate copies sent around the world
-Then, in 1960:
$1 \mathrm{~m}=1650763.73 \times$ wavelength ${ }^{86} \mathrm{Kr}$ (orange)
- 1983 until now (strict definition):
$1 \mathrm{~m}=$ distance light travels in $1 /(299792458) \mathrm{sec}$

Measurement
Lifetime of the proton (predicted)
Age of the universe
Age of the pyramid of Cheops
Human life expectancy
Length of a day
Time between human heartbeats
Lifetime of the muon
Shortest lab light pulse
Lifetime of the most unstable particle
The Planck time ${ }^{a}$

Time Interval in Seconds
$1 \times 10^{39}$
$5 \times 10^{17}$
$1 \times 10^{11}$
$2 \times 10^{9}$
$9 \times 10^{4}$
$8 \times 10^{-1}$
$2 \times 10^{-6}$ $6 \times 10^{-15}$
$1 \times 10^{-23}$
$1 \times 10^{-43}$

Some standards used through the ages:
-Length of the day
-Period of vibration of a quartz crystal
-Now we use atomic clocks
${ }^{a}$ This is the earliest time after the big bang at which the laws of physics as we know them can be applied.
1 second equivalent to 9192631770 oscillations of the light emitted by a cesium-133 atom ( ${ }^{133} \mathrm{Cs}$ ) at a specified wavelength (adopted 1967)

## United States Naval Observatory

## Mass

| Object | Mass in Kilograms | - Kilogram standard is a Pt-Ir cylinder in Paris |
| :---: | :---: | :---: |
| Known universe | $1 \times 10^{53}$ |  |
| Our galaxy | $2 \times 10^{41}$ | - Accurate copies have been sent around the world; the US version is housed in a vault at NIST |
| Sun | $2 \times 10^{30}$ |  |
| Moon | $7 \times 10^{22}$ |  |
| Asteroid Eros | $5 \times 10^{15}$ |  |
| Small mountain | $1 \times 10^{12}$ | A second mass standard: |
| Ocean liner | $7 \times 10^{7}$ | -The ${ }^{12} \mathrm{C}$ atom has been assigned a mass of 12 atomic mass units (u) |
| Elephant | $5 \times 10^{3}$ |  |
| Grape | $3 \times 10^{-3}$ |  |
| Speck of dust | $7 \times 10^{-10}$ |  |
| Penicillin molecule | $5 \times 10^{-17}$ | $1 u=1.6605402 \times 10$ |
| Uranium atom | $4 \times 10^{-25}$ |  |
| Proton | $2 \times 10^{-27}$ | -The masses of all other atoms |
| Electron | $9 \times 10^{-31}$ | e determined relative to ${ }^{12} \mathrm{C}$ |

## Chapter 1 Summary



Luminosity: candela (cd)

Numbers are often written with prefixes or in scientific notation to express powers of 10 . Accuracy is shown by the number of significant figures:


## Ch.2: Motion in one-dimension

-We will define the position of an object using the variable $x$, which measures the position of the object relative to some reference point (origin) along a straight line (x-axis).







## Average velocity and speed

$$
v_{\text {avg }}=\bar{v}=\frac{\Delta x}{\Delta t}=\frac{x_{2}-x_{1}}{t_{2}-t_{1}}
$$

- Like displacement, the sign of $v_{a v g}$ indicates direction Average speed $s_{\text {avg }}$ :

$$
s_{\text {avg }}=\bar{s}=\frac{\text { total distance }}{\Delta t}
$$

- $s_{\text {avg }}$ does not specify a direction; it is a scalar as opposed to a vector \&, thus, lacks an algebraic sign - How do $\boldsymbol{v}_{\text {avg }}$ and $s_{\text {avg }}$ differ?



## Instantaneous velocity and speed



## Acceleration

- An object is accelerating if its velocity is changing

Average acceleration $a_{a v g}$ :

$$
a_{a v g}=\bar{a}=\frac{\Delta v}{\Delta t}=\frac{v_{2}-v_{1}}{t_{2}-t_{1}}
$$

Instantaneous acceleration a:

$$
a=\lim _{\Delta t \rightarrow 0} \frac{\Delta v}{\Delta t}=\frac{d v}{d t}=\frac{d}{d t}\left(\frac{d x}{d t}\right)=\frac{d^{2} x}{d t^{2}}
$$

- This is the second derivative of the $\boldsymbol{x}$ vs. $t$ graph
- Like $\boldsymbol{x}$ and $\boldsymbol{v}$, acceleration is a vector
- Note: direction of $a$ need not be the same as $v$





## Summarizing

Displacement:

$$
\Delta x=x_{2}-x_{1}
$$

Average velocity: $\quad v_{a v g}=\bar{v}=\frac{\Delta x}{\Delta t}=\frac{x_{2}-x_{1}}{t_{2}-t_{1}}$
Average speed:

$$
s_{a v g}=\bar{s}=\frac{\text { total distance }}{\Delta t}
$$

Instantaneous velocity:

$$
v=\frac{d x}{d t}=\text { local slope of } x \text { versus } t \text { graph }
$$

Instantaneous speed: magnitude of $v$

## Summarizing

Average acceleration: $\quad a_{a v g}=\bar{a}=\frac{\Delta v}{\Delta t}=\frac{v_{2}-v_{1}}{t_{2}-t_{1}}$
Instantaneous acceleration:

$$
a=\frac{d v}{d t}=\text { local slope of } v \text { versus } t \text { graph }
$$

In addition:

$$
a=\frac{d}{d t}\left(\frac{d x}{d t}\right)=\frac{d^{2} x}{d t^{2}}=\text { curvature of } x \text { versus } t \text { graph }
$$

SI units for $a$ are $\mathrm{m} / \mathrm{s}^{2}$ or $\mathrm{m} . \mathrm{s}^{-2}$ ( $\mathrm{ft} / \mathrm{min}^{2}$ also works)

